# Diazepines and Naphthyridines from the Rearrangement of Thienoindolizidinone Oximes

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Thienoindolizidinone oximes previously reported [1] were treated with polyphosphoric acid and led to thienopyrrolidinodiazepines or thienonaphthyridines. The rearrangement depends on the structural form of the starting oxime and the results obtained are discussed.

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In a preceeding paper [1] we synthesized several ketones by intramolecular Friedel-Crafts acylation of 5-oxo-1-(thienylmethyl)prolines. We wish to report herein the Beckmann rearrangement of the oximes 1a-d corresponding to these ketones.

In high acidic medium two types of reaction can be observed: the Beckmann rearrangement leading to cyclic lactams 2a-d or the fragmentation of the oxopyrrolidine ring leading to naphthyridines 3b-d (see Scheme II).

As previously described [1], oximes  $\mathbf{1a-d}$  may exist in two forms: configuration  $\mathbf{A}$  in which the OH group of the oxime is syn to the thiophene ring and configuration  $\mathbf{B}$  in which the OH group is anti.

Previous investigations [2,3] have elucidated that the oxime 4 (isomer B) gave only the naphthyridine 5. Thus it appeared of interest to study oximes annelated to a thiophene ring.

Therefore, when oximes 1a-d are heated with polyphosphoric acid at 120° under nitrogen we formed with compound 1a the diazepine 2a and with compounds 1b-d a mixture of diazepines and naphthyridines 3b-d. All these reactions gave a quantitative yield of the crude products. The results, summarized in Table 1, are in direct correspondance with the configuration of the starting oxime.

Table 1 Composition\* of starting oximes and lactams after reaction of 1 with PPA. Oximes Products (%) \_OH A = 100% 100% 2a За 1a и~Он A = 40%; B = 60% 35% 65% 2b 3b 1b N~OH A = 81%; B = 19% 80% 20% 2c 3с B = 100% 58% 2d 3d 1d

Determined in the crude product by <sup>1</sup>H nmr spectroscopy (200 MHz).

Oxime 1a (A) afforded exclusively the diazepine 2a according to the normal Beckmann rearrangement. The structure of 2a is supported by its <sup>1</sup>H nmr spectrum (Table 2). The multiplet at 5.11-5.20 ppm is assigned to proton  $H_{8a}$ . It shows a significant chemical shift (+0.7 ppm) from the corresponding ketone [1] ( $\delta H_{8a} = 4.50$  ppm) because it is influenced by the two nitrogen atoms of the diazepine ring. We also observe that the -N-CH<sub>2</sub>- protons of the di-

### Scheme II

azepine ring appear as an AB quartet with chemical shifts of 4.42 and 4.83 ppm and a coupling constant  $J=17.8~\mathrm{Hz}$  characteristic of methylene protons. We have already observed a non-equivalence of the methylene protons in benzothienothiazocines [4] and thienoindolizidines [1]. Furthermore a  $^{13}\mathrm{C}$  nmr spectral evaluation (Table 3) confirms the structure 2a. Carbon  $C_{3a}$  is typically shifted upfield in the diazepine ( $\delta=139.4~\mathrm{ppm}$ ) compared to the corresponding ketone ( $\delta=149.5~\mathrm{ppm}$ ).

Under the same conditions (polyphosphoric acid, 120°) the mixture of oximes 1b (A = 40%, B = 60%) led to a mixture of pyrrolo[1,2-a]thieno[3,2-e][1,3]diazepine 2b (35%) and naphthyridine 3b (65%). The oxime 1b (A) afforded, as in the preceeding case, the Beckmann product. In contrast oxime 1b (B) underwent a fragmentation of the oxopyrrolidine ring with formation of the thieno[2,3-c]-[1,5]naphthyridine 3b. These two compounds were easily separated by chromatography on silica gel. The 1H, 13C nmr and ir data (Tables 2 and 3) of 2b are similar to those of 2a. For both, we notice constant chemical shifts characteristic of the carbon atoms of the diazepine ring ( $C_{8a}$  = 65.9 ppm,  $C_{3a} = 139.4$  ppm). The diazepine corresponding to a migration of the thiophene ring to the nitrogen atom was not found (Beckmann product from 1b (B)) probably due to the fragile character of the lactam linkage of the oxopyrrolidine ring. Compound 3b was characterized by its H<sub>9</sub> proton  $\delta = 8.72$  ppm as a singlet signal. These observations confirm the results obtained elsewhere [2] with oxime 4 (B) which afforded, under the same conditions, exclusively the naphthyridine 5.

 $Table \ 2 \\ 1H NMR Chemical Shifts of Pyrrolothieno[1,3] diazepines \textbf{2a-d} in DMSO-d_6, \delta (ppm)$ 

Compound No.	(CH <sub>2</sub> ) <sub>3</sub>	${ m H_{8a}}$	${ m H_4}$	NH	H <sub>arom</sub>	J,Hz
2a	1.98-2.27 m	5.11-5.20 m	4.83 d, 4.42 d	8.68 s	7.75 d (H <sub>2</sub> ), 7.05 d (H <sub>3</sub> )	$5.1(\mathrm{H_2},\mathrm{H_3}),17.8(\mathrm{H_4gem})$
2ь	1.95-2.57 m	5.10-5.21 m	4.75 d, 4.63 d	8.77 d	$7.49 d (H_2), 7.29 d (H_1)$	$5.2~(\mathrm{H_1,H_2}),16.7~(\mathrm{H_4~gem})$
<b>2</b> e	2.00-2.62 m	5.32-5.45 m	5.23 d, 4.72 d	9.05 в	8.03 d (1H), 7.94 d (1H) 7.50 m (2H)	7.8 (H <sub>arom</sub> ), 7.5 (H <sub>arom</sub> ) 18.4 (H <sub>4</sub> gem)
<b>2d</b> [a]	1.88-2.58 m	5.03-5.15 m	4.74 d, 4.31 d	8.94 d		15.9 (H <sub>4</sub> gem)

[a] 2.21 (s, 3H, CH<sub>3</sub>).

 $Table \ 3$   $^{13}C\ NMR\ Chemical\ Shifts\ of\ Pyrrolothieno[1,3] diazepines\ \textbf{2a-d}\ in\ DMSO-d_6,\ \delta\ (ppm)$ 

Compound No.	C <sub>8</sub>	$C_7$	C4	C <sub>8a</sub>	C <sub>3a</sub>	$C_3$	$C_2$	$c_1$	$C_{10a}$	C <sub>10</sub>	C <sub>6</sub>
2a	24.7	28.7	42.6	65.9	139.4	128.7	131.7	=	134.6	164.1	173.4
2b	23.8	29.6	40.4	65.9	139.5	_	124.7	129.4	135.1	165.7	172.6
2c [a]	25.0	28.6	41.6	65.8	134.8	137.9	139.1	_	133.8	164.4	173.7
2d [b]	22.9	30.1	39.0	65.8	135.1	_	122.5	135.6	134.4	165.4	171.9

[a] 122.6, 123.2, 124.9, 127.1 ( $C_{arom}$ ). [b] 12.9 ( $CH_3$ ).

cedure.

The results obtained from oxime 1d are interesting. This one, exclusively with configuration **B** (100%), gave a mixture of naphthyrine 3d (58%) and diazepine 2d (42%). According to the preceeding results and to the literature cited [2] the naphthyridine is the normal rearrangement product. The formation of diazepine 2d can be explained by an initial isomerization of the oxime in acidic medium followed by the Beckmann rearrangement. This isomerization has been previously reported elsewhere [5] during the formation of benzothienoazocine. On the other hand, we did not observe the formation of a nitrile product as in the benzene series [2]. Compounds 2d and 3d were also separated by chromatography and easily identified from their spectral data (ir, <sup>1</sup>H and <sup>13</sup>C nmr). There is a perfect analogy of the chemical shift of carbons C3a and C8a respectively  $\delta = 135.1$  and 65.8 ppm, with those of diazepines 2a and 2b (see Tables 2 and 3).

From these results it was of interest to study an oxime annelated to a benzothiophene system. Thus, oxime 1c was allowed to react with polyphosphoric acid at 120° to afford a mixture of diazepine and naphthyridine. Oxime 1c (A) gave the expected diazepine 2c (80%) and oxime 1c (B) gave naphthyridine 3c (20%). These percentages are in agreement with those of starting products 1c (A) = 81% and 1c(B) = 19%. As previously, both compounds 2c and 3c were separated by chromatography on silica gel (R<sub>f</sub> values are given in the Experimental). The spectral data are summarized in Tables 2 and 3. As with the oxime la (A), oxime lc (A) afforded exclusively the diazepine 2c corresponding to the Beckmann rearrangement and as oxime 4 (B), oxime 1c (B) led to the naphthyridine 3c. We did not observe an isomerization product or a nitrile product.

In conclusion, it seems that the intramolecular interaction between the hydrogen of the hydroxyl group and the sulfur atom would control the regioselectivity of the Beckmann rearrangement. Thus it is a convenient route to the new tetrahydropyrrolothienodiazepine system. Further investigations with a pyrrolidine or a piperidine ring instead of oxopyrrolidine are in progress and the results will be published soon.

## **EXPERIMENTAL**

Melting points are uncorrected. The infrared spectra of solids (potassium bromide) were recorded on a Beckman IR 20 spectrometer. The nmr spectra were recorded on a Bruker AC-200 spectrometer (200 MHz) in DMSO-d<sub>6</sub> using tetramethylsilane ('H) or DMSO-d<sub>6</sub> (1<sup>3</sup>C, δ = 39.5 ppm) as the internal standards. Ascending thin layer chromatography was performed on precoated plates of silica gel 60 F 254 (Merck) and the spots visualized using an ultraviolet lamp or iodine vapor. E. Merck silica gel 60 F (70-300 mesh) was used for column chromatography. The elemental analyses were carried out by the microanalysis laboratory of INSA at Rouen, F 76130 M<sup>T</sup>. S<sup>T</sup>. Aignan. The starting oximes

**1a-d** were prepared according to a reported procedure [1]. Reaction of Oximes **1a-d** with Polyphosphoric Acid. General Pro-

Finely powdered oxime **la-d** (5 mmoles) was added to hot (120°) stirred polyphosphoric acid (20 g) within 5 minutes. The mixture was stirred under nitrogen for 30 minutes. The hot solution was decanted over crushed ice (200 ml) and the resulting mixture was basified to pH 8 with 50% sodium hydroxide solution. Extraction with chloroform (3 x 50 ml), followed by the usual work up of the organic layer, gave a solid residue. The crude

A. The less soluble heterocycles **2a** and **3b** were obtained by crystallization from ethanol. The mother liquor was concentrated in vacuo.

product was processed according to one of the following proced-

B. The mixture of diazepines **2b-d** and naphthyridines **3b-d** were separated by column chromatography on silica gel (100 g for 1 g of crude product) eluting with chloroform/acetone (8:2). The thieno[c][1,5]naphthyridines **3b-d** were obtained as the first eluted products. Further elution afforded the pyrrolothienodiazepines **2b-d**. The R<sub>F</sub> values (chloroform/acetone = 8/2) are: 0.21 (**2a**), 0.22 (**2b**), 0.29 (**2c**), 0.34 (**2d**), 0.36 (**3b**), 0.43 (**3c**), and 0.58 (**3d**).

4,7,8,8a-Tetrahydropyrrolo[1,2-a]thieno[2,3-e][1,3]diazepine-6,10-(9H)-dione (**2a**).

This compound was obtained in 63% yield, mp 273-275° (from ethanol); ir: 1674 (CO), 1700 (CO) cm<sup>-1</sup>.

Anal. Calcd. for  $C_{10}H_{10}N_2O_2S$ : C, 54.03; H, 4.54; N, 12.61. Found: C, 53.87; H, 4.14; N, 12.42.

5a,6,7,10-Tetrahydropyrrolo[1,2-a]thieno[3,2-e][1,3]diazepine-4(5H),8-dione (**2b**).

This compound was obtained in 42% yield, mp 210-211° (from benzene-cyclohexane); ir: 1644 (CO), 1704 (CO) cm<sup>-1</sup>.

Anal. Calcd. for  $C_{10}H_{10}N_2O_2S$ : C, 54.03; H, 4.54; N, 12.61. Found: C, 53.83; H, 4.42; N, 12.47.

1,2,5,12a-Tetrahydropyrrolo[1,2-a]benzo[b]thieno[2,3-e][1,3]diazepine-3,11(12H)-dione (2c).

This compound was obtained in 45% yield, mp 263-264° (from ethanol); ir: 1645 (CO), 1710 (CO) cm $^{-1}$ .

Anal. Calcd. for  $C_{14}H_{12}N_2O_2S$ : C, 61.74; H, 4.45; N, 10.29. Found: C, 61.42; H, 4.17; N, 10.10.

2-Chloro-3-methyl-5a,6,7,10-tetrahydropyrrolo[1,2-a]thieno[3,2-e][1,3]diazepine-4(5H),8-dione (2d).

This compound was obtained in 18% yield, mp 226-227° (from ethanol); ir: 1645 (CO), 1698 (CO) cm<sup>-1</sup>.

Anal. Calcd. for  $C_{11}H_{11}N_2O_2SCI$ : C, 48.79; H, 4.10; N, 10.35. Found: C, 48.53; H, 3.98; N, 10.13.

The <sup>1</sup>H and <sup>13</sup>C nmr spectra of diazepines **2a-d** are given in Tables 2 and 3.

6,7-Dihydrothieno[2,3-c][1,5]naphthyridin-5(4H)-one (3b).

This compound was obtained in 68% yield, mp 268-270° (from ethanol); ir: 1684 (CO) cm<sup>-1</sup>; <sup>1</sup>H nmr:  $\delta$  2.62 (t, 2 H, CH<sub>2</sub>CO), 3.09 (t, 2 H, CH<sub>2</sub>, J = 8.2 Hz), 7.82 (d, 1 H, H<sub>3</sub>, J = 5.5 Hz), 7.97 (d, 1 H, H<sub>2</sub>), 8.72 (s, 1 H, H<sub>9</sub>), 10.53 (br s, 1 H, NH).

Anal. Calcd. for  $C_{10}H_8N_2OS$ : C, 58.80; H, 3.96; N, 13.72. Found: C, 58.53; H, 3.93; N, 13.45.

3,4-Dihydrobenzo[b]thieno[3,2-c][1,5]naphthyridin-2(1H)-one (3c).

This compound was obtained in 36% yield, mp >280°; ir: 1668 (CO) cm $^{-1}$ ;  $^{1}\text{H}$  nmr:  $\delta$  2.72 (t, 2 H, CH $_{2}\text{CO}$ ), 3.22 (t, 2 H, CH $_{2}$ , J = 8.0 Hz), 7.50-7.62 (m, 2 H, H $_{8}$  and H $_{9}$ ), 8.10-8.19 (m, 1 H, H $_{7}$ ), 8.40-8.48 (m, 1 H, H $_{10}$ ), 9.16 (s, 1 H, H $_{6}$ ), 10.66 (br s, 1 H, NH). Anal. Calcd. for  $C_{14}H_{10}N_{2}\text{OS}$ : C, 66.11; H, 3.97; N, 11.02. Found: C, 66.03; H, 3.90; N, 11.02.

2-Chloro-6,7-dihydro-3-methylthieno[2,3-c][1,5]naphthyridin-5(4H)-one (3d).

This compound was obtained in a yield of 36%, mp 219-221° (from ethanol); ir: 1650 (CO) cm<sup>-1</sup>; <sup>1</sup>H nmr:  $\delta$  2.58 (s, 3 H, CH<sub>3</sub>), 2.64 (t, 2 H, CH<sub>2</sub>CO), 3.11 (t, 2 H, CH<sub>2</sub>, J = 8.0 Hz), 8.69 (s, 1 H, H<sub>9</sub>), 10.66 (br s, 1 H, NH).

Anal. Calcd. for  $C_{11}H_9N_2OSCl$ : C, 52.57; H, 3.60; N, 11.09. Found: C, 52.37; H, 3.50; N, 10.96.

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